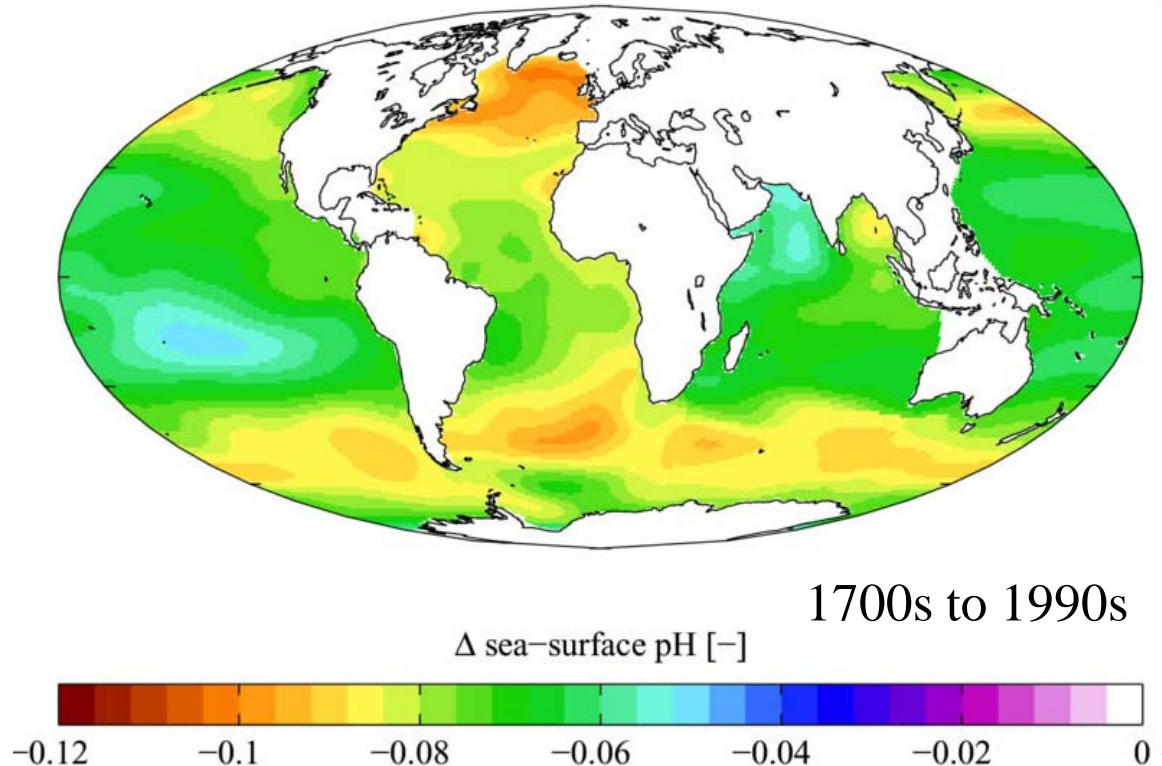
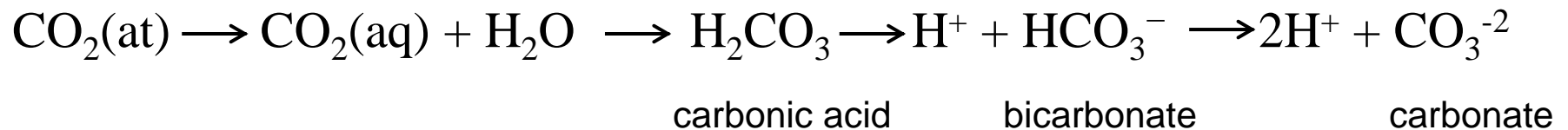
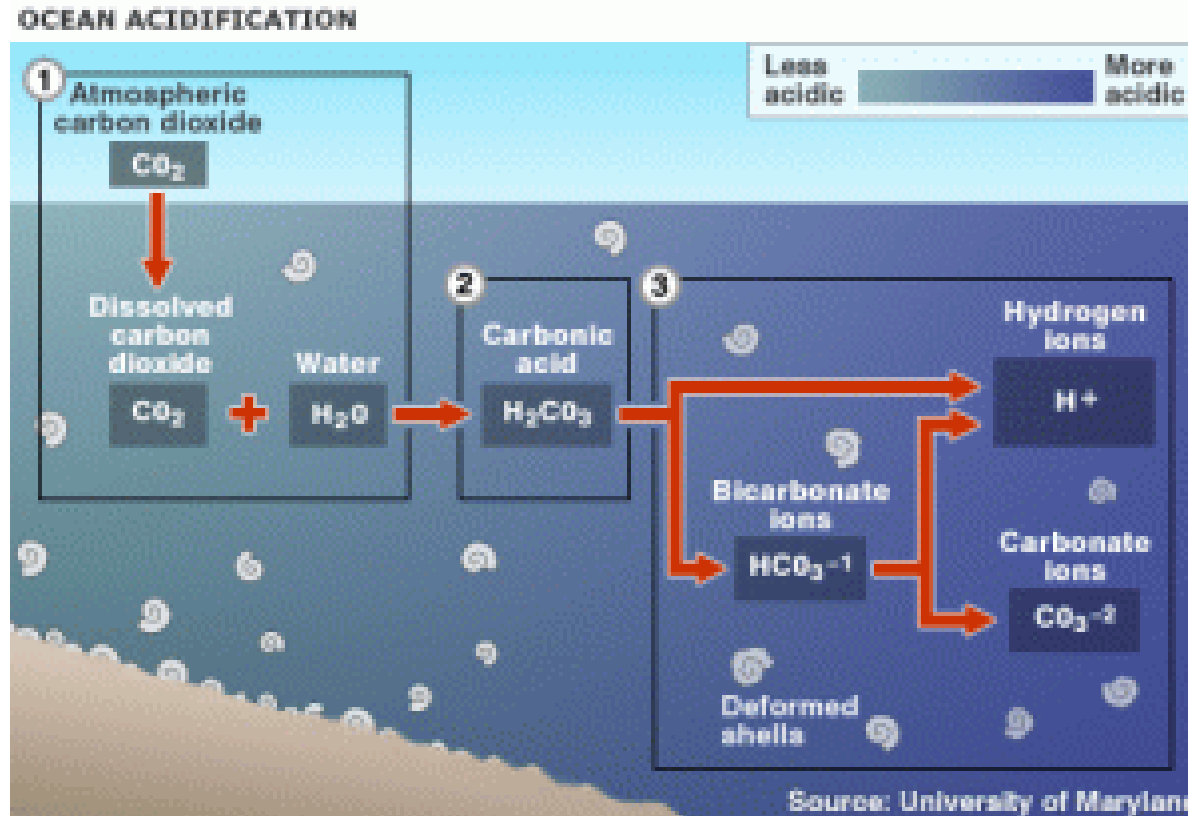


Impacts of Ocean Acidification (Hypercapnia)

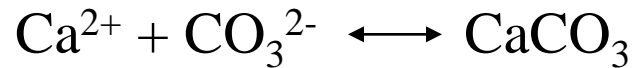


LO: extrapolate effects of changing pH conditions to ELH stages of marine fish species

Ocean Carbonate Chemistry



Carbonate and Biology



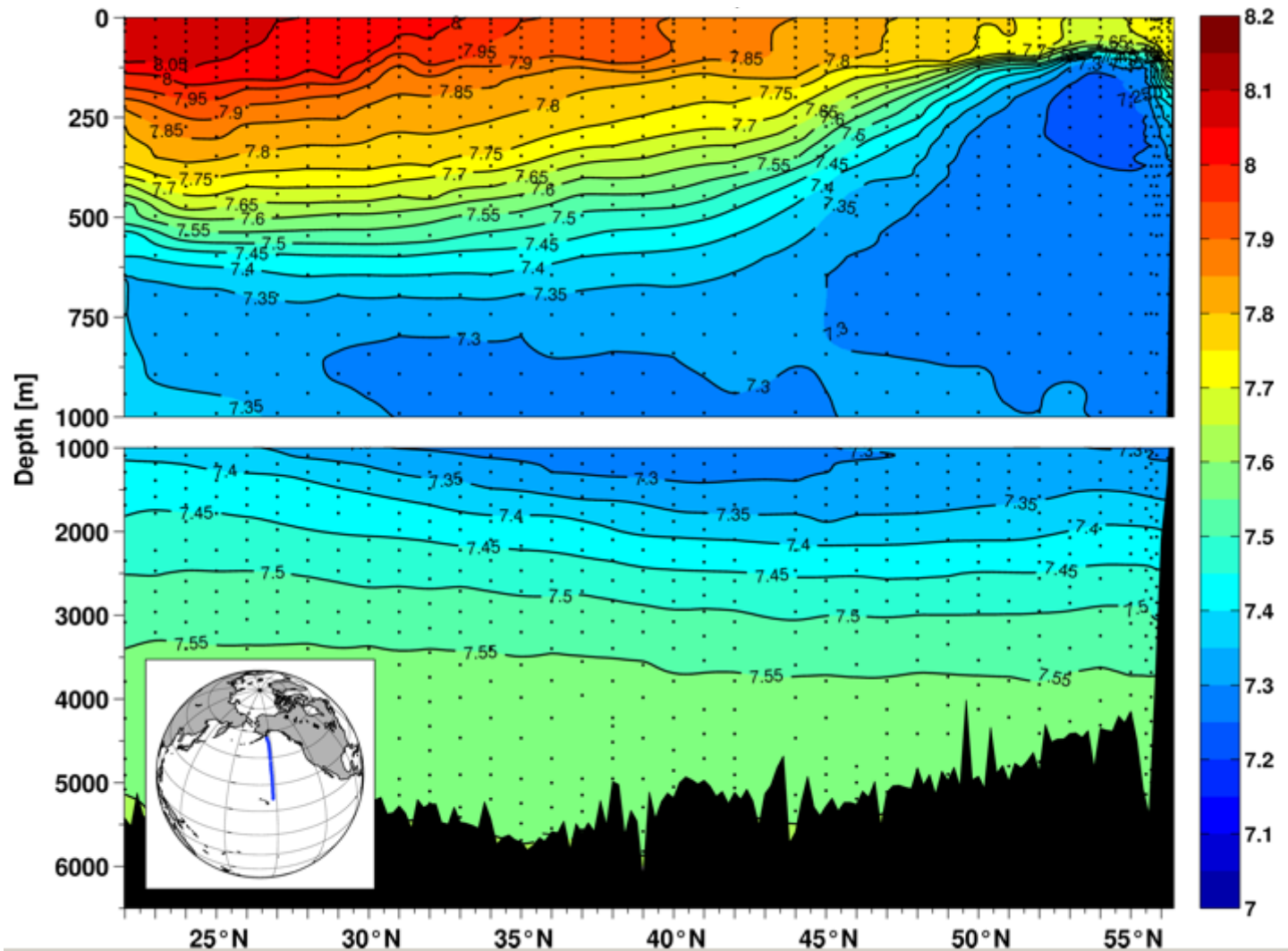
Carbonate ions are combined with Calcium to form Calcium Carbonate

$$\text{pH} = -\log [\text{H}^+]$$

1/3 of $\text{CO}_2(\text{at})$ absorbed by ocean, open ocean pH decreased over past 200 years by ~ 0.1 units (Caldeira and Wickett 2003)

biologically important below pH 7.8

North Pacific pH

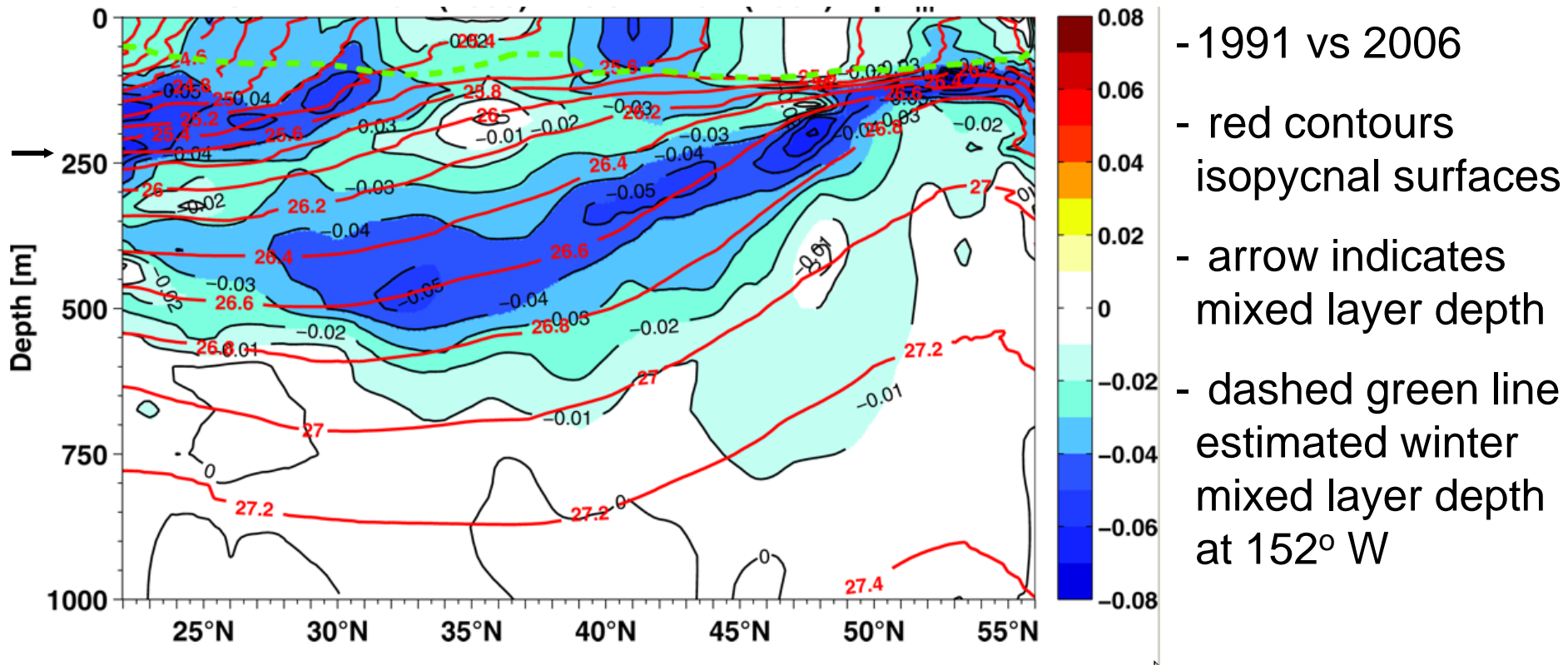


- Pacific 10% higher than Atlantic due to respiration products

- cruise in 2006

- long 152° W

Change in North Pacific pH

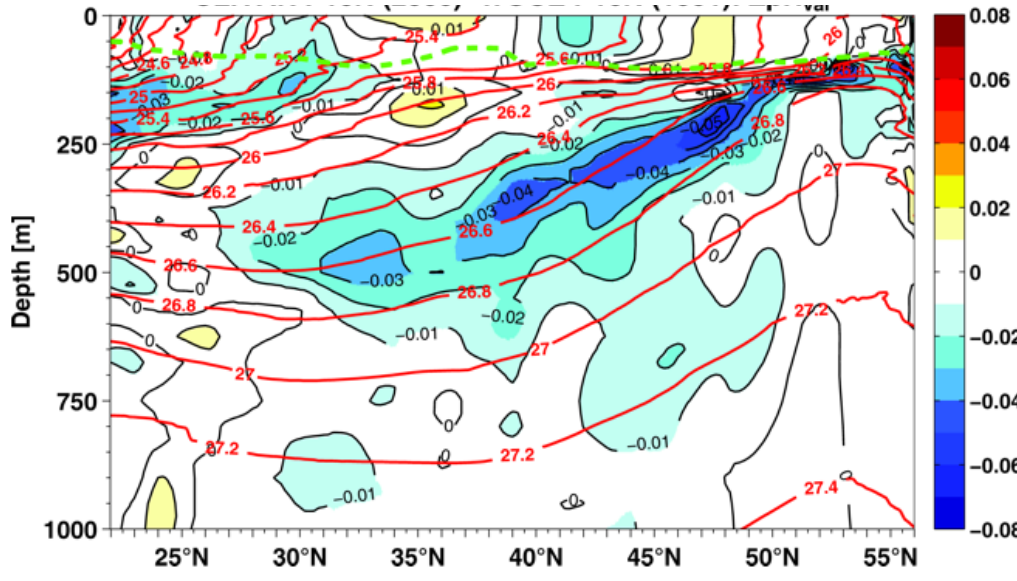


Generally decline but no significant changes

Overall average change over lat and depth: - 0.023

Between surface and 150 m: - 0.03

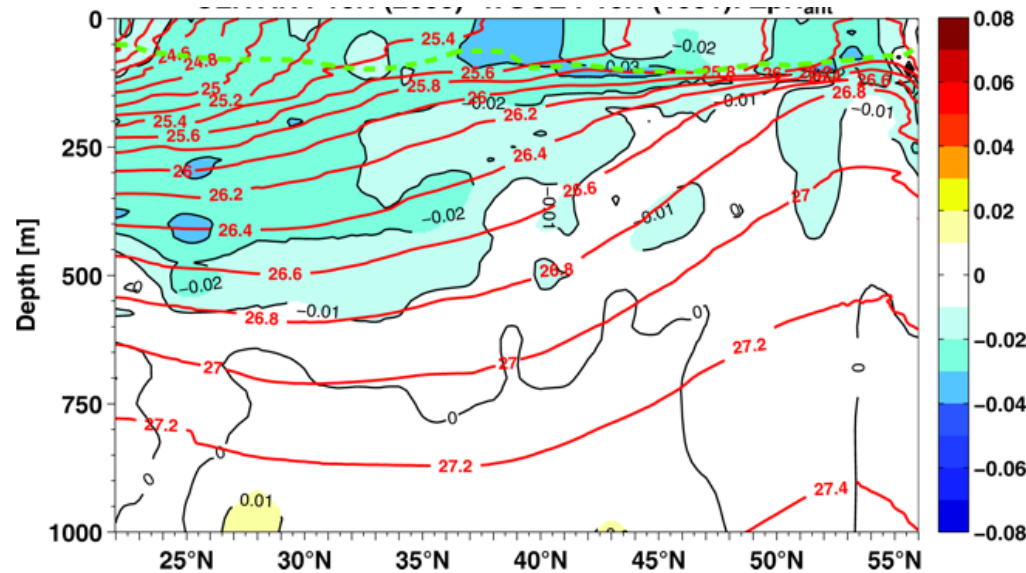
Sources of Carbon




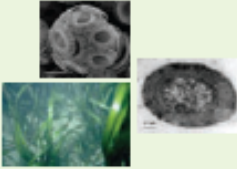


pH change attributed to natural variability

- acidification \approx CO₂(at)

pH change attributed to anthropogenic sources



Non-Fish Biological Impacts

Physiological response	Major group	Species studied	Response to increasing CO ₂			
			a	b	c	d
Calcification 	Coccolithophores ¹	4	2	1	1	1
	Planktonic Foraminifera	2	2	-	-	-
	Molluscs	4	4	-	-	-
	Echinoderms ¹	3	2	1	-	-
	Tropical corals	11	11	-	-	-
	Coralline red algae	1	1	-	-	-
Photosynthesis² 	Coccolithophores ³	2	-	2	2	-
	Prokaryotes	2	-	-	1	-
	Seagrasses	5	-	-	-	-
Nitrogen Fixation 	Cyanobacteria	1	-	1	-	-
Reproduction 	Molluscs	4	4	-	-	-
	Echinoderms	1	1	-	-	-

- tipping point for calcareous species below 7.8 pH

1) Increased calcification had substantial physiological cost; 2) Strong interactive effects with nutrient and trace metal availability, light, and temperature; 3) Under nutrient replete conditions.

Undersaturation of Aragonite



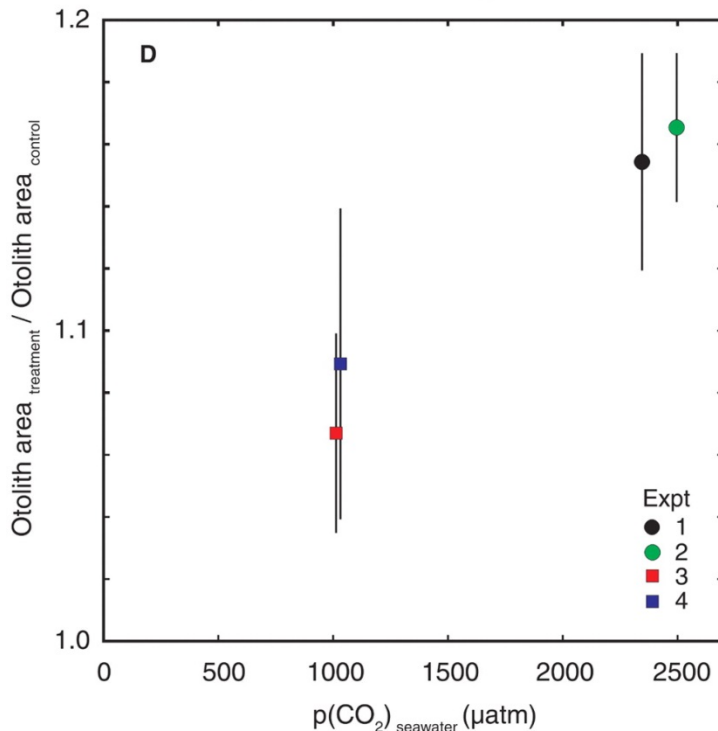
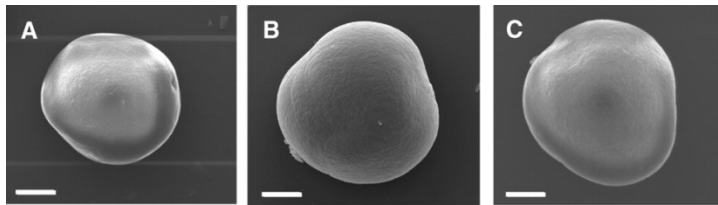
pteropod

- surface waters in high latitudes predicted undersaturated by 2050
- pteropods prey for zooplankton & fish
- pink salmon in N. Pacific >60% diet
- pteropod range may contract to shallower depths & lower latitudes
- potential to reorganize planktonic and benthic communities

pH Effect on Otoliths

Otoliths: aragonite-protein bilayers

Doney et al. (2009): decreased pH slows calcification



White sea bass

(*Atractoscion nobilis*)

A=430 (control)

B=1000

C=2500 μatm p(CO₂) seawater

Increased area and mass of otoliths

B: 7-9% area, 10-14% heavier

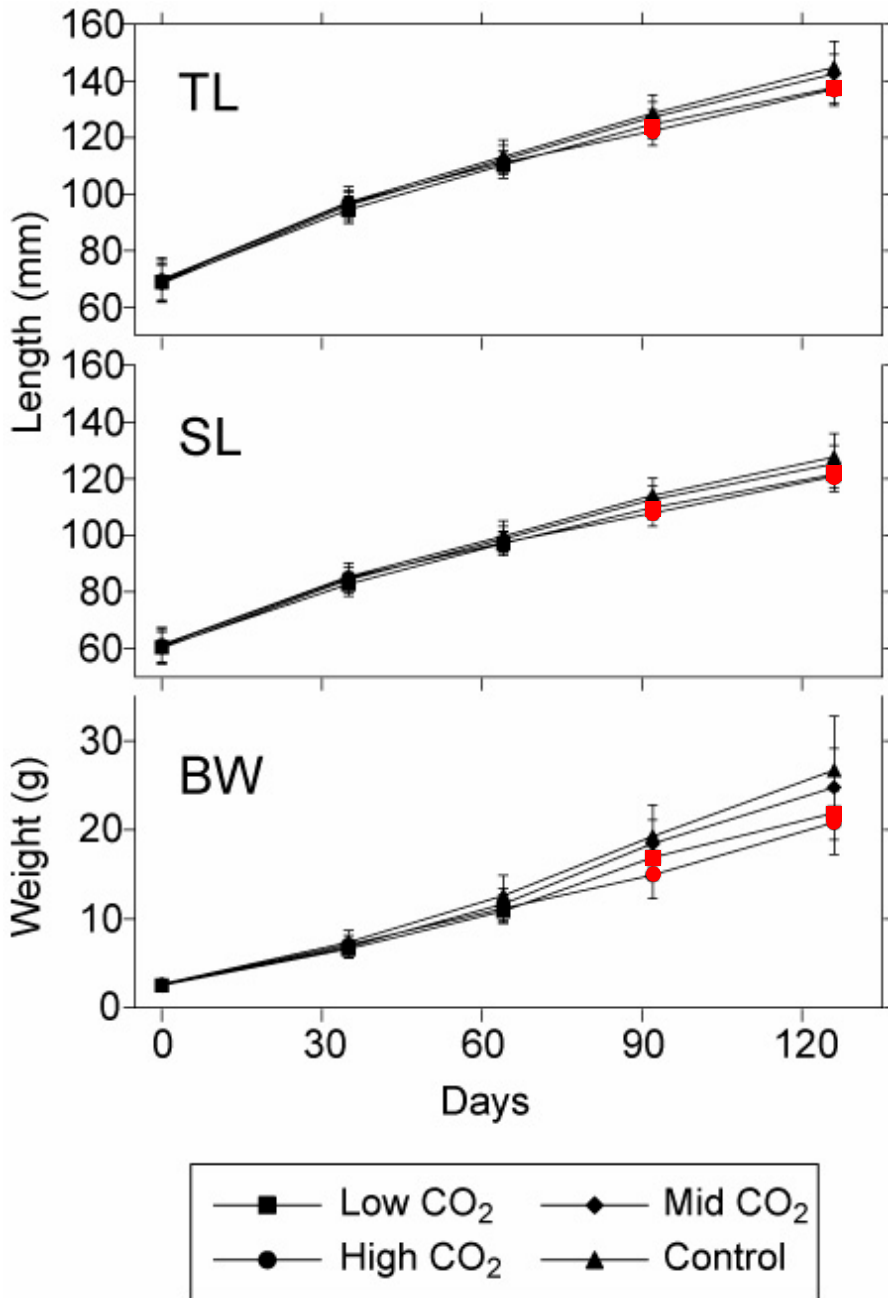
C: 15-17% area, 24-26% heavier

Dry mass did not vary

Control H⁺ and Ca⁺ but not CO₂ around otolith, which increases CO₃²⁻

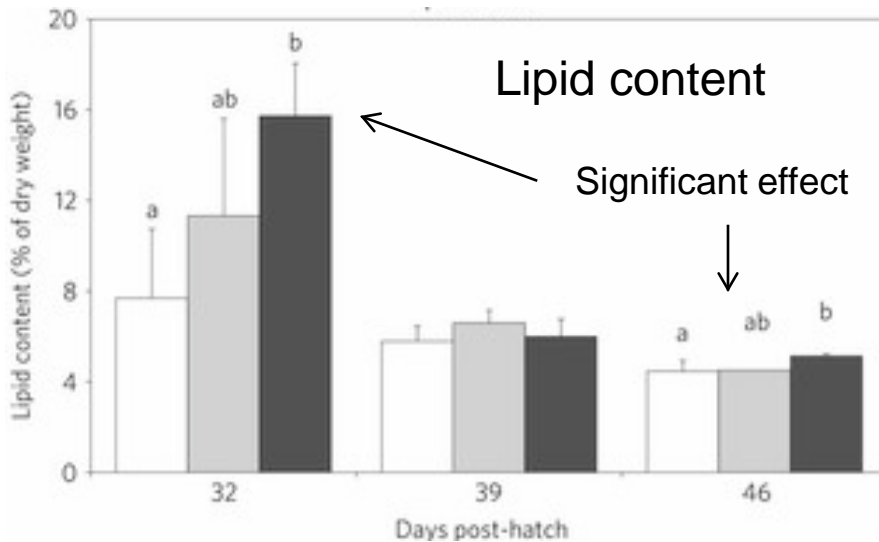
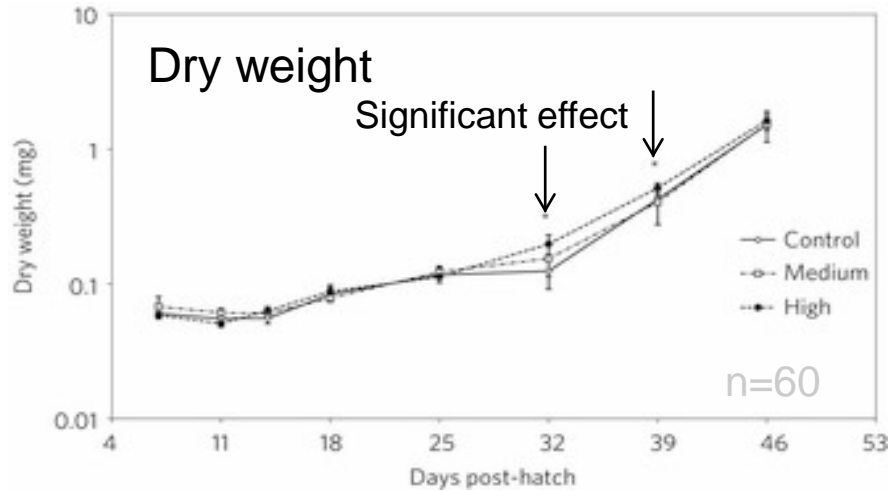
Checkley et al. 2009

Fish Growth

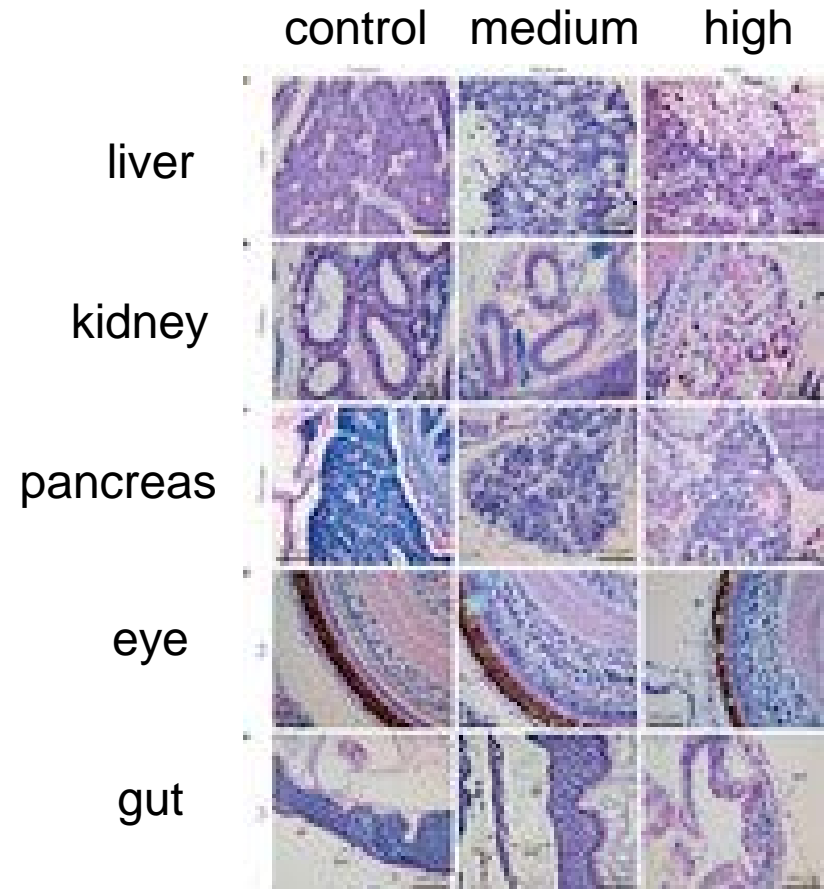


- Japanese sillago (*Sillago sihama*) whiting
- growth significantly reduced in 90 days (red symbols)

Larval Fish Tissue Damage



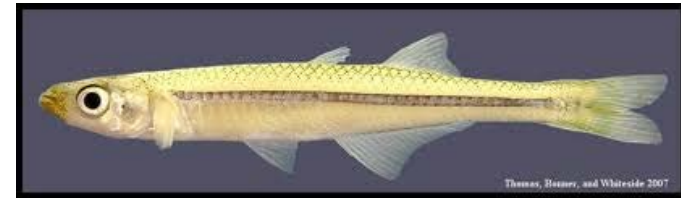
Atlantic cod (*Gadus morhua*)



380; 1,800; 4,200 $\mu\text{atm pCO}_2$

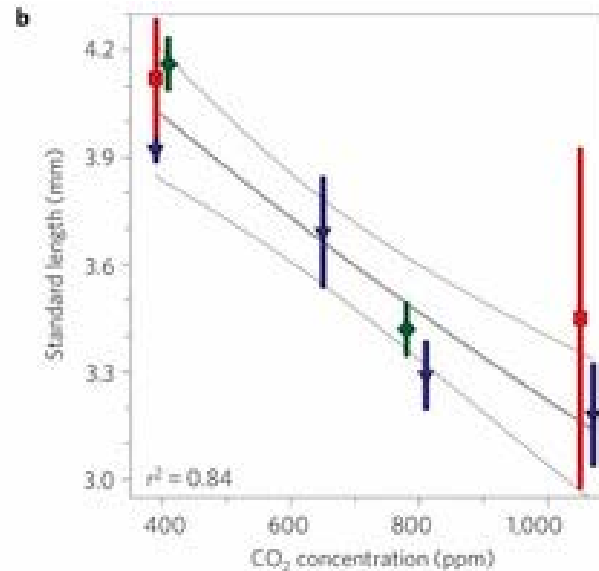
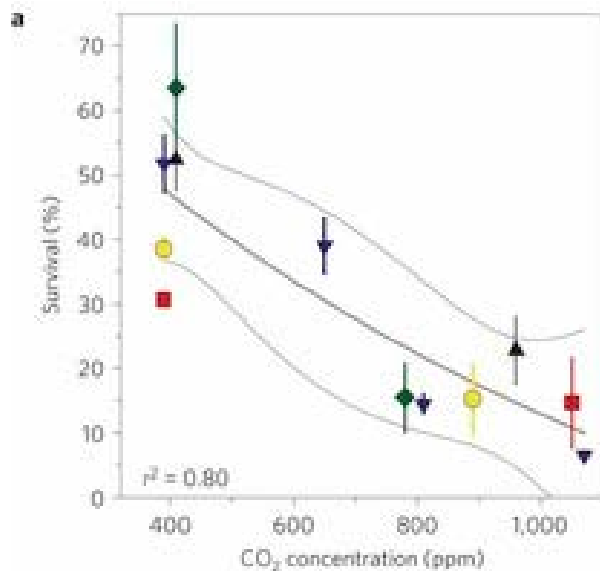
Frommel et al. 2011

Fish Reduced Survival



Embryos – 1 week post hatch survival

Inland silverside
(*Menidia beryllina*)

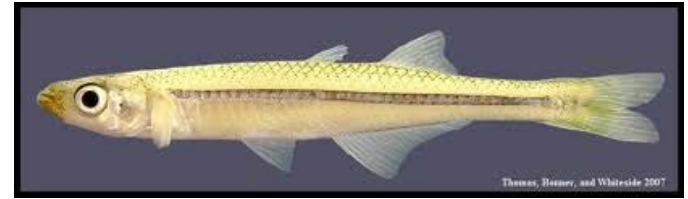


↑
now

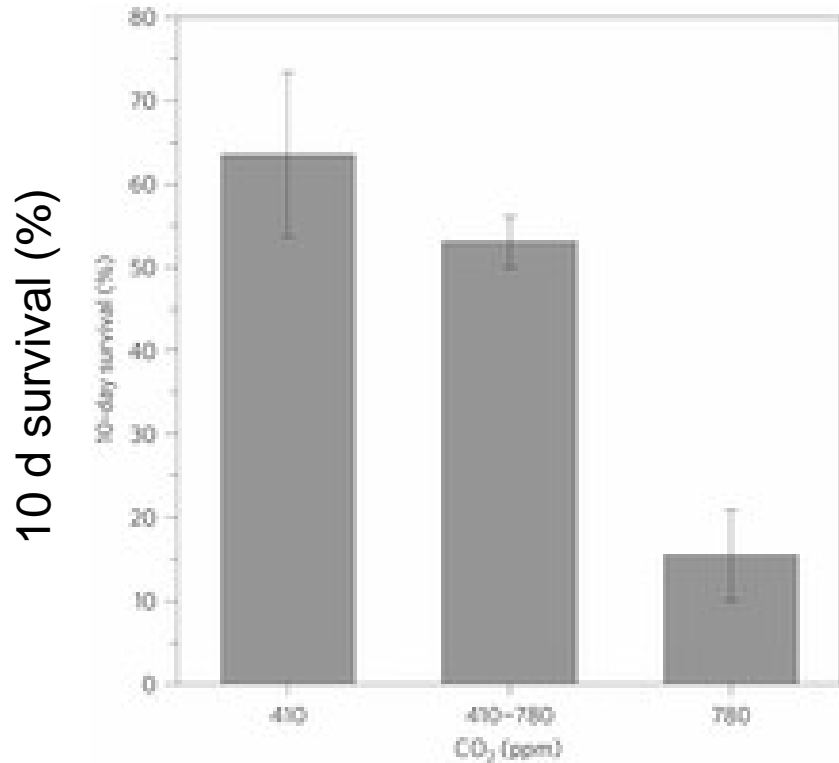
↑
mid

↑
end
century

Fish Reduced Survival



Inland silverside
(*Menidia beryllina*)

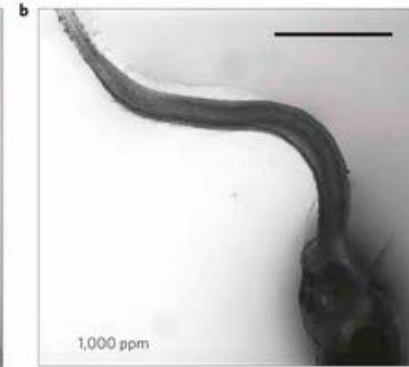


CO₂ (ppm)

increased
after 5 d
post
fertilization



400 ppm

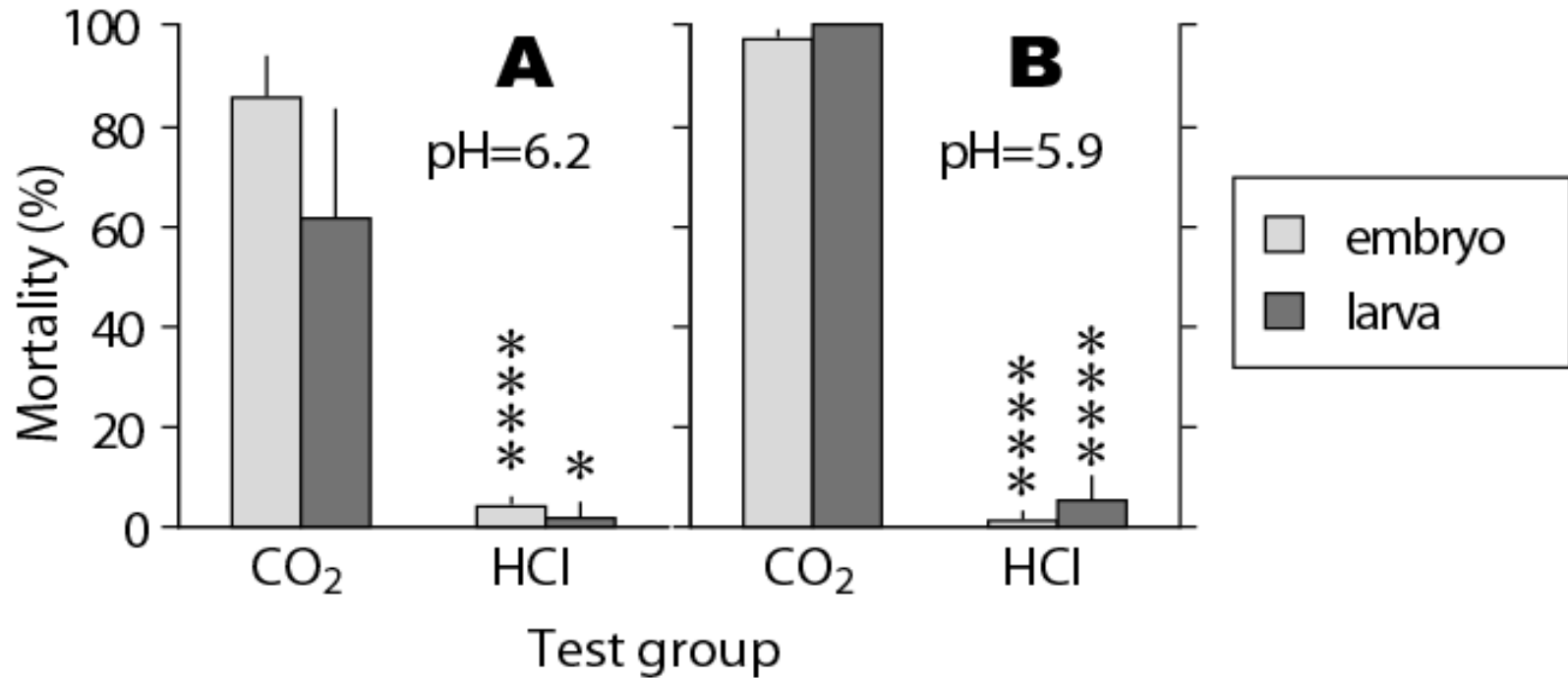


800 ppm



1000 ppm

Fish Mortality Cause: CO₂ vs H⁺

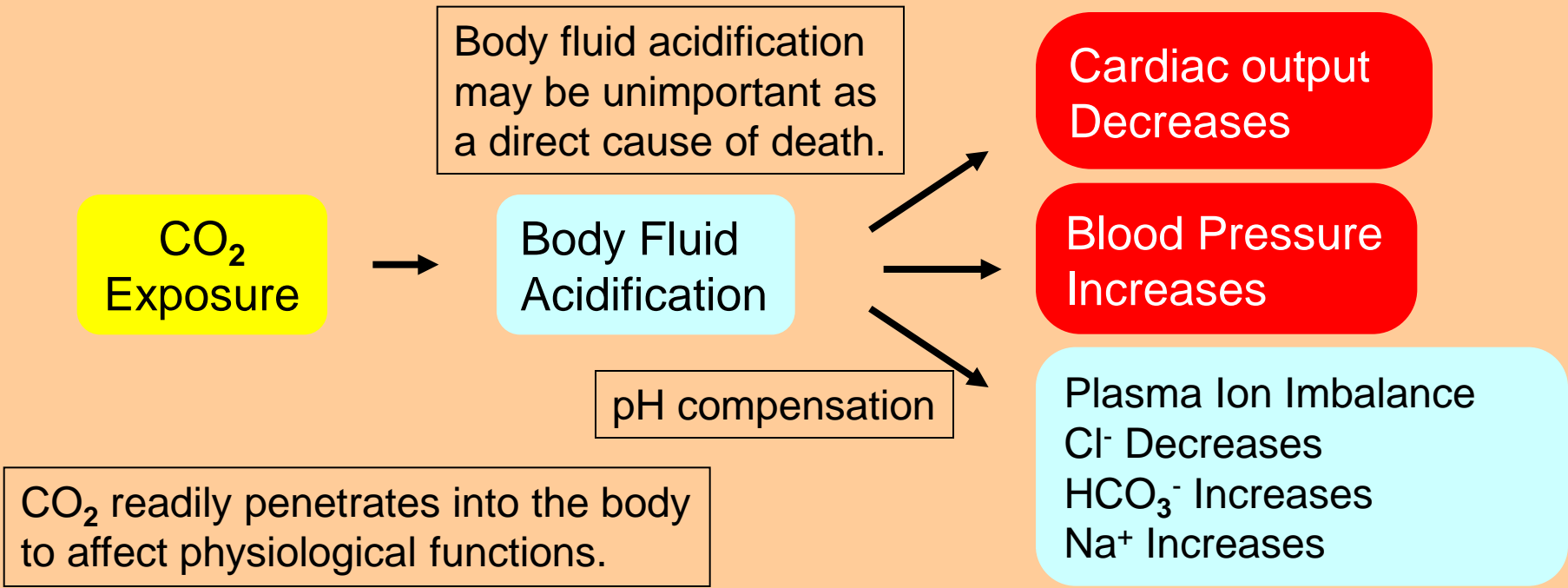
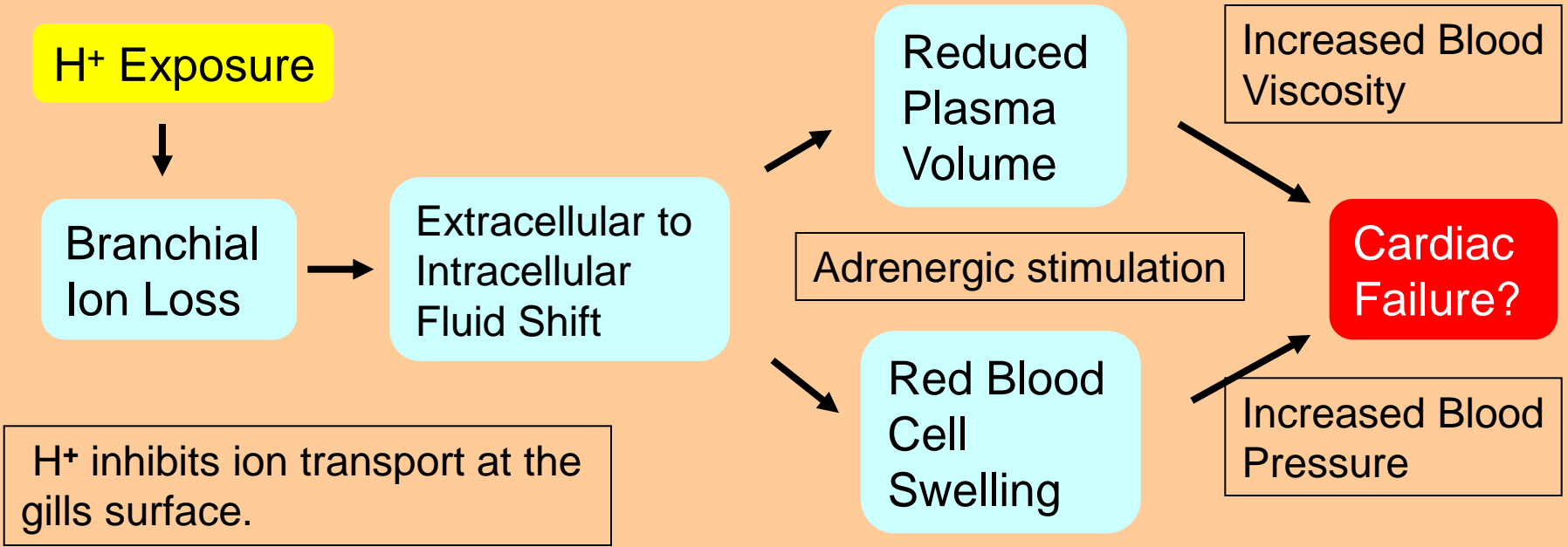


Lethal effect of CO₂ and acid on embryos (N = 5) and larvae (N = 3) of silver seabream at two pH conditions. Exposure period: embryo 360 min, larva 24 h
*Significant difference between groups.

Kikkawa et al. (2004): Marine Pollution Bulletin 48, 108.

Mortality Due to CO₂ and H⁺

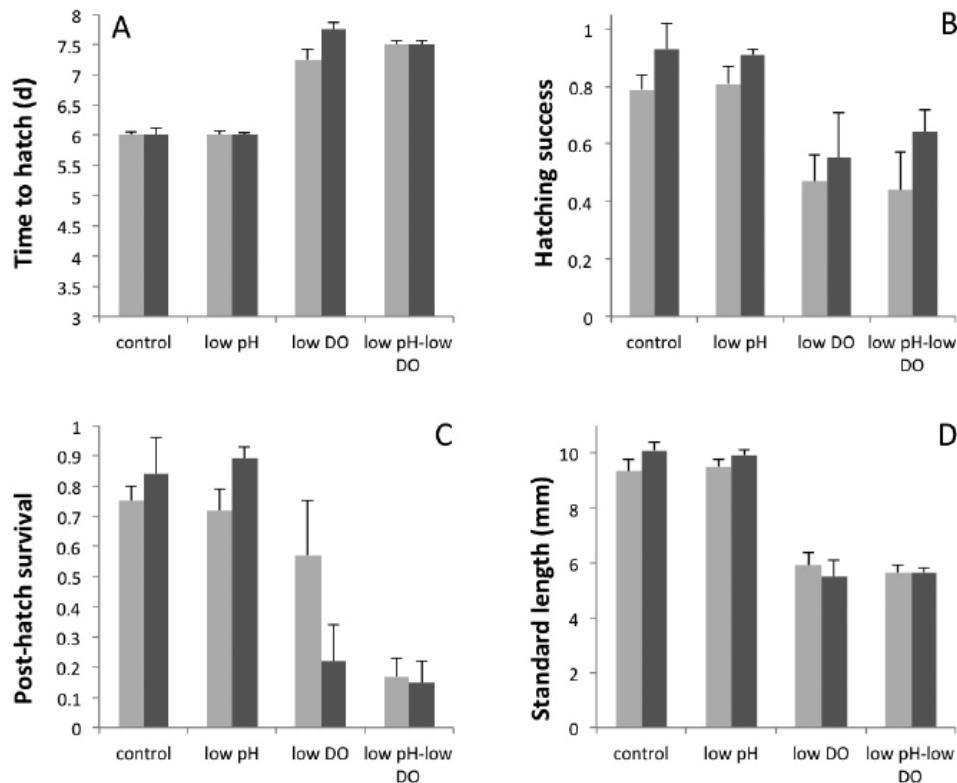
- Physiological responses to CO₂ and acids are different.
- CO₂ diffuses into body, and acidifies body fluid of both intracellular and extracellular compartments. Fish kill mechanism by high CO₂ is not fully understood.
- Acid exposure inhibits active ion transport across gills, and increased passive ion movements. Cause of fish kill by acid exposure is thought **not** to be blood acidification but cardiac failure by hemoconcentration.



Combined Effects of pH and O₂

Forage fish: Silverside (*Menidia beryllina*, *Menidia menidia*); Sheepshead minnow (*Cyprinodon variegatus*)

- egg through early larval, May & June experiments



Low pH differed little from the control, but when combined with low DO, had a significant effect (dependent on variable and species)

Other Effects

Effect	Stage	Cause	Reference
mercury toxicity	larval	mercury vs selenium	MacDonald et al. 2015 Metallomics 7: 1247-1255 MacDonald et al. 2015 J. inorganic Biochem 151: 10-17
Spinal deformities, hatching delay, mortality	larval	heavy metals	Sfakianakis et al. 2015 Environental Research 137: 246-255
endocrine disruptors	juveniles	gene expression for contolling homoe balance	Liao et al. J. Hazardous Materials 277:141-149 Chu et al. Chemosphere 152:181-189
lost orientation due to loss of hearing	settling larva	CO2 on hearing (assume enlarged otolith development)	Rossi et al. 2016 Biology Letters 12: 20150937
noisier ocean	all	reduced noise absorption	Hester et al. 2008. Geophys. Res. Letters 35: L19601

Summary Effects

Factor	ELH stage & response	Management
Increased acidity (decrease in pH) H ⁺ and Carbonate	<i>Adult spawning:</i>	
	<i>Egg stage:</i>	
	<i>Early larvae:</i>	
	<i>Juvenile:</i>	
Temperature		